

SET ~ 3

Series Z1XYW/1

प्रश्न-पत्र कोड
Q.P. Code 31/1/3

रोल नं.

Roll No.

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परीक्षार्थी प्रश्न-पत्र कोड को उत्तर-पुस्तिका के मुख-पृष्ठ पर अवश्य लिखें।
Candidates must write the Q.P. Code on the title page of the answer-book.

- कृपया जाँच कर लें कि इस प्रश्न-पत्र में मुद्रित पृष्ठ 31 हैं।
- प्रश्न-पत्र में दाहिने हाथ की ओर दिए गए प्रश्न-पत्र कोड को परीक्षार्थी उत्तर-पुस्तिका के मुख-पृष्ठ पर लिखें।
- कृपया जाँच कर लें कि इस प्रश्न-पत्र में 39 प्रश्न हैं।
- कृपया प्रश्न का उत्तर लिखना शुरू करने से पहले, उत्तर-पुस्तिका में प्रश्न का क्रमांक अवश्य लिखें।
- इस प्रश्न-पत्र को पढ़ने के लिए 15 मिनट का समय दिया गया है। प्रश्न-पत्र का वितरण पूर्वाह्न में 10.15 बजे किया जाएगा। 10.15 बजे से 10.30 बजे तक परीक्षार्थी केवल प्रश्न-पत्र को पढ़ेंगे और इस अवधि के दौरान वे उत्तर-पुस्तिका पर कोई उत्तर नहीं लिखेंगे।
- Please check that this question paper contains 31 printed pages.
- Q.P. Code given on the right hand side of the question paper should be written on the title page of the answer-book by the candidate.
- Please check that this question paper contains 39 questions.
- Please write down the serial number of the question in the answer-book before attempting it.
- 15 minute time has been allotted to read this question paper. The question paper will be distributed at 10.15 a.m. From 10.15 a.m. to 10.30 a.m., the candidates will read the question paper only and will not write any answer on the answer-book during this period.

विज्ञान

SCIENCE

निर्धारित समय : 3 घण्टे

Time allowed : 3 hours

अधिकतम अंक : 80

Maximum Marks : 80

31/1/3

106 C

— 1 —

P.T.O.

General Instructions :

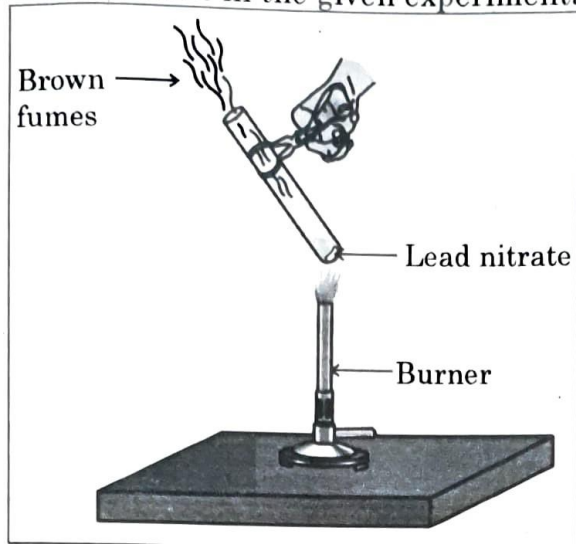
Read the following instructions carefully and strictly follow them :

- (i) This question paper consists of **39** questions. All questions are compulsory.
- (ii) Question paper is divided into **FIVE** sections viz. Section **A, B, C, D** and **E**.
- (iii) In Section **A** - question number **1 to 20** are Multiple Choice Questions (MCQs) carrying **1** mark each.
- (iv) In Section **B** - question number **21 to 26** are Very Short Answer (VSA) type questions carrying **2** marks each. Answer to these questions should be in the range of **30 to 50** words.
- (v) In Section **C** - question number **27 to 33** are Short Answer (SA) type questions carrying **3** marks each. Answer to these questions should be in the range of **50 to 80** words.
- (vi) In Section **D** - question number **34 to 36** are Long Answer (LA) type questions carrying **5** marks each. Answer to these questions should be in the range of **80 to 120** words.
- (vii) In Section **E** - question number **37 to 39** are of 3 source-based/case-based units of assessment carrying **4** marks each with sub-parts.
- (viii) There is no overall choice. However, an internal choice has been provided in some Sections.

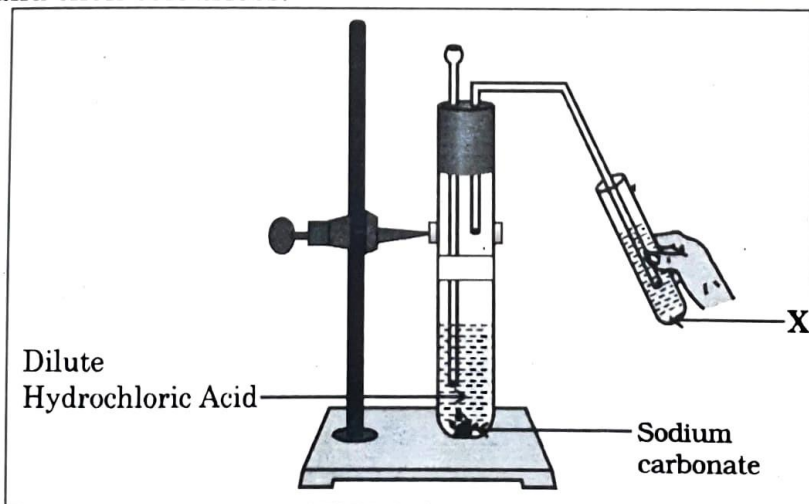
SECTION - A

Select and write one most appropriate option out of the four options given for each of the questions 1 - 20 :

1. The emission of brown fumes in the given experimental set-up is due to 1

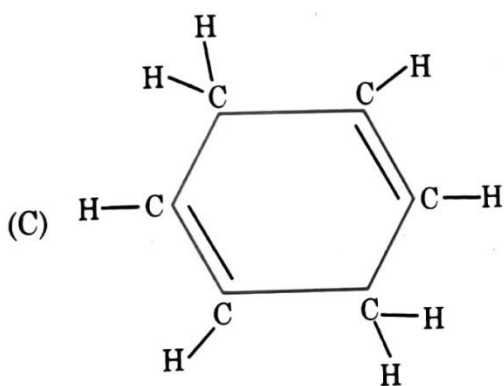
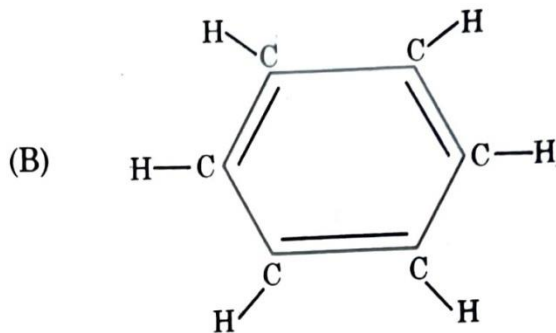
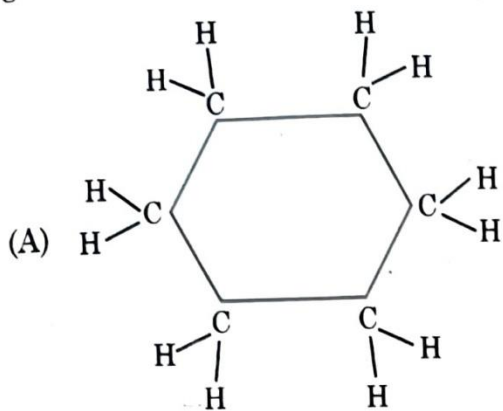


- (a) thermal decomposition of lead nitrate which produces brown fumes of nitrogen dioxide.
(b) thermal decomposition of lead nitrate which produces brown fumes of lead oxide.
(c) oxidation of lead nitrate forming lead oxide and nitrogen dioxide.
(d) oxidation of lead nitrate forming lead oxide and oxygen.
2. In the experimental setup given below, it is observed that on passing the gas produced in the reaction in the solution 'X' the solution 'X' first turns milky and then colourless. 1



The option that justifies the above stated observation is that 'X' is aqueous calcium hydroxide and

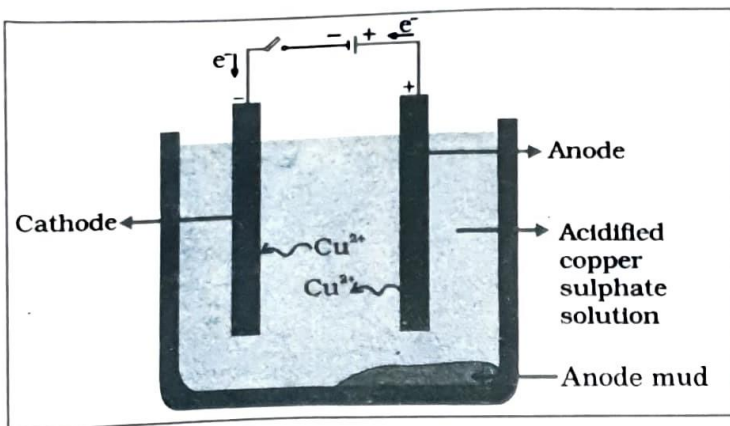
6. Consider the structures of the three cyclic carbon compounds A, B and C given below and select the correct option from the following : 1



- (a) A and C are isomers of hexane and B is benzene.
- (b) A is an isomer of hexane, B is benzene and C is an isomer of hexene.
- (c) A is a saturated cyclic hydrocarbon and B and C are unsaturated cyclic hydrocarbons.
- (d) A is cyclohexane and B and C are the isomers of benzene.

7. The following diagram shows the electrolytic refining of copper :

1



Which of the following statements is **incorrect** description of the process ?

- (a) The impure metal from the anode dissolves into the electrolyte.
- (b) The pure metal from the electrolyte is deposited on the cathode.
- (c) Insoluble impurities settle down at the bottom of the anode.
- (d) On passing the current through the electrolyte, the pure metal from the anode dissolves into the electrolyte.

8. Consider the following statements about small intestine and select the one which is **NOT** correct :

1

- (a) The length of the small intestine in animals differs as it depends on the type of food they eat.
- (b) The small intestine is the site of complete digestion of food.
- (c) The small intestine receives secretions from liver and pancreas.
- (d) The villi of the small intestine absorb water from the unabsorbed food before it gets removed from the body via the anus.

9. An organism which breaks down the food material outside the body and then absorbs it is

1

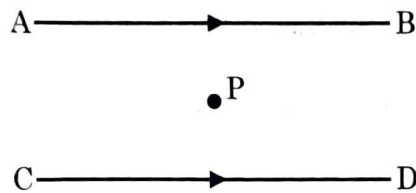
- (a) a plant parasite, Cuscuta
- (b) an animal parasite, Tapeworm
- (c) a bacteria, Rhizobium
- (d) a fungi, Rhizopus

10. Which of the following metals **do not** corrode in moist air ? 1
- (a) Copper
 - (b) Iron
 - (c) Gold
 - (d) Silver
11. Select from the following the correct statement about tropic movement in plants : 1
- (a) It is due to stimulus of touch and temperature.
 - (b) It does not depend upon the direction of stimulus received.
 - (c) It is observed only in roots and not in stems.
 - (d) It is a growth related movement.
12. The statement that correctly describes the characteristic(s) of a gene is : 1
- (a) In individuals of a given species, a specific gene is located on a particular chromosome.
 - (b) A gene is not the information source for making proteins in the cell.
 - (c) Each chromosome has only one gene located all along its length.
 - (d) All the inherited traits in human beings are not controlled by genes.
13. In domestic electric circuits the wiring with 15 A current rating is for the electric devices which have 1
- (a) higher power ratings such as geyser.
 - (b) lower power ratings such as fan.
 - (c) metallic bodies and low power ratings.
 - (d) non-metallic bodies and low power ratings.

14. If four identical resistors, of resistance 8 ohm, are first connected in series so as to give an effective resistance R_s , and then connected in parallel so as to give an effective resistance R_p , then the ratio $\frac{R_s}{R_p}$ is 1

- (a) 32 (b) 2
(c) 0.5 (d) 16

15. The resultant magnetic field at point 'P' situated midway between two parallel wires (placed horizontally) each carrying a steady current I is 1



- (a) in the same direction as the current in the wires.
(b) in the vertically upward direction.
(c) zero
(d) in the vertically downward direction.
16. If the current I through a resistor is increased by 100% (at constant temperature), the increase in power dissipated will be 1
- (a) 100%
(b) 200%
(c) 300%
(d) 400%

Q. No. 17 to 20 are Assertion – Reason based questions.

These consist of two statements – Assertion (A) and Reason (R). Answer these questions selecting the appropriate option given below :

- (a) Both (A) and (R) are true and (R) is the correct explanation of (A).
- (b) Both (A) and (R) are true, but (R) is not the correct explanation of (A).
- (c) (A) is true, but (R) is false.
- (d) (A) is false, but (R) is true.

17. **Assertion (A) :** Genes inherited from the parents decide the sex of a child. 1

Reason (R) : X chromosome in a male child is inherited from his father.

18. **Assertion (A) :** The colour of aqueous solution of copper sulphate turns colourless when a piece of lead is added to it. 1

Reason (R) : Lead is more reactive than copper, and hence displaces copper from its salt solution.

19. **Assertion (A) :** The strength of the magnetic field produced at the centre of a current carrying circular coil increases on increasing the number of turns in it. 1

Reason (R) : The current in each circular turn has the same direction and the magnetic field due to each turn then just adds up.

20. **Assertion (A) :** Left atrium receives oxygenated blood from pulmonary vein. 1

Reason (R) : Right atrium transfers deoxygenated blood to the right ventricle, which pumps it to the lungs for oxygenation.

SECTION - B

Q. No. 21 to 26 are very short answer questions.

21. A plant hormone helps in bending of stem towards light. Name the hormone and explain how is the movement achieved. 2

22. (a) (i) A compound 'X' which is prepared from gypsum has the property of hardening when mixed with proper quantity of water. 2

Identify 'X' and write its chemical formula.

(ii) State the difference in chemical composition between baking soda and baking powder.

OR

(b) Write balanced chemical equation for the reaction that occurs when : 2

(i) blue coloured copper sulphate crystals are heated and

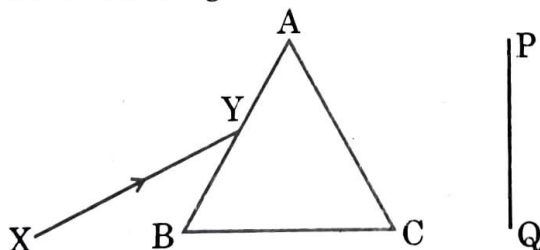
(ii) Sodium hydrogen carbonate is heated during cooking.

23. Name the part of the human excretory system where nephrons are found. Write the structure and function of nephrons. 2

24. Write the constituents of solder alloy. Which property of solder makes it suitable for welding electrical wires ? 2

25. Why is damage to the ozone layer a cause for concern ? What steps are being taken to limit this damage ? 2

26. (a) A narrow beam XY of white light is passing through a glass prism ABC as shown in the diagram : 2



Trace it on your answer sheet and show the path of the emergent beam as observed on the screen PQ.

Name the phenomenon observed and state its cause.

OR

(b) It is observed that the power of an eye to see nearby objects as well as far off objects diminishes with age. 2

(i) Give reason for the above statement.

(ii) Name the defect that is likely to arise in the eyes in such a condition.

(iii) Draw a labelled ray diagram to show the type of corrective lens used for restoring the vision of such an eye.

SECTION – C

Q. No. 27 to 33 are short answer questions.

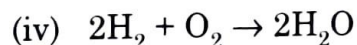
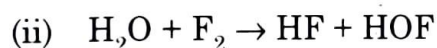
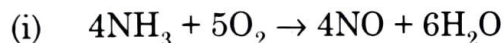
27. A chemical compound X is used in glass, soap and paper industries. On treatment with ethanoic acid, it forms salt, water and carbon dioxide. 3

(a) Identify X.

(b) How is this compound obtained from brine ? Write the equations involved.

(c) State the number of molecules of water of crystallization present in compound X.

28. (a) Identify the reducing agent in the following reactions : 3



(b) Define a redox reaction in terms of gain or loss of oxygen.

29. A student wants to project the image of a candle flame on a screen 60 cm in front of a mirror by keeping the candle at a distance of 15 cm from its pole. 3

(a) Name the type of mirror used.

(b) Also calculate :

(i) Magnification of the image produced

(ii) Distance between object and its image

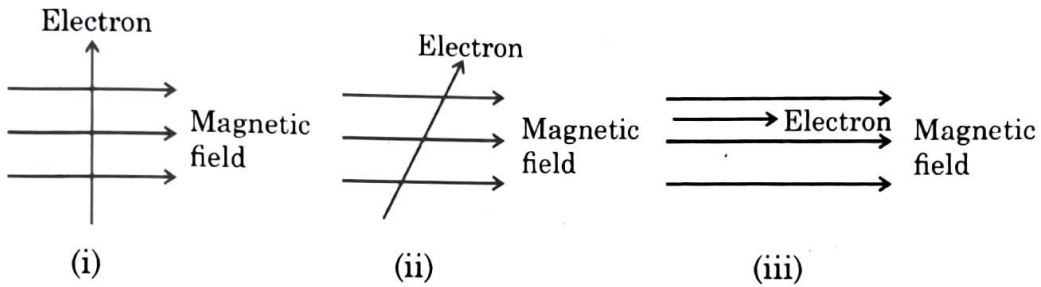
(c) Draw a ray diagram to show the image formation.

30. (a) (i) State the role of ATP in cellular respiration. 3
 (ii) What ensures sufficient exchange of gases in plants ?
 (iii) State the conditions on which the direction of diffusion of gases in plant depend upon.

OR

- (b) (i) What is the internal energy reserve in plants and animals ? 3
 (ii) How desert plants perform photosynthesis if their stomata remain closed during the day ?

31. (a) (i) State the rule used to find the force acting on a current carrying conductor placed in a magnetic field. 3
 (ii) Given below are three diagrams showing entry of an electron in a magnetic field. Identify the case in which the force will be (1) maximum and (2) minimum respectively. Give reason for your answer.



OR

- (b) (i) Draw the pattern of magnetic field lines of 3
 (1) a current carrying solenoid
 (2) a bar magnet
 (ii) List two distinguishing features between the two fields.
32. (a) (i) Why does a kitchen garden called an artificial ecosystem while a forest is considered to be a natural ecosystem ? 3
 (ii) While designing an artificial ecosystem at home, write any two things to be kept in mind to convert it into a self-sustaining system. Give reason to justify your answer.

OR

- (b) (i) Construct a food chain of four trophic levels comprising the following : 3
Hawk, snake, plants, rat.
- (ii) 20,000 J of energy was transferred by the producers to the organism of second trophic level. Calculate the amount of energy that will be transferred by organisms of the third trophic level to the organisms of the fourth trophic level.
33. A person cannot see distinctly the object placed beyond 5 m from his eyes. Name the defect of vision the person is suffering from. Draw a ray diagram to illustrate this defect. List its two possible causes. Name the lens used for the correction of this defect. 3

SECTION - D

Q. No. 34 to 36 are long answer questions.

34. (a) Mention the role of the following organs of human male reproductive system : 5
(i) Testis
(ii) Scrotum
(iii) Vas-deferens
(iv) Seminal vesicle
- (b) What is Placenta ? State its function in a human female.
35. (a) An electric iron consumes energy at a rate of 880 W when heating is at the maximum rate and 330 W when the heating is at the minimum. If the source voltage is 220 V, calculate the current and resistance in each case. 5
- (b) What is heating effect of electric current ?
- (c) Find an expression for the amount of heat produced when a current passes through a resistor for some time.
36. (a) A saturated organic compound 'A' belongs to the homologous series of alcohols. 5
On heating 'A' with concentrated sulphuric acid at 443 K, it forms an unsaturated compound 'B' with molecular mass 28 u.
The compound 'B' on addition of one mole of hydrogen in the presence of Nickel, changes to a saturated hydrocarbon 'C'.

- (i) Identify A, B and C.
- (ii) Write the chemical equations showing the conversion of A into B.
- (iii) What happens when compound C undergoes combustion ?
- (iv) State one industrial application of hydrogenation reaction.
- (v) Name the products formed when compound A reacts with sodium.

OR

- (b) (i) With the help of diagram, show the formation of micelles, when soap is applied on oily dirt. 5
- (ii) Take two test tubes X and Y with 10 mL of hard water in each. In test tube 'X', add few drops of soap solution and in test tube 'Y' add a few drops of detergent solution. Shake both the test tubes for the same period.
 - (1) In which test tube the formation of foam will be more ? Why ?
 - (2) In which test tube is a curdy solid formed ? Why ?

SECTION - E

Q. No. 37 to 39 are case based / data based questions with 2 to 3 short sub-parts. Internal choice is provided in one of these sub-parts.

37. In order to trace the inheritance of traits Mendel crossed pea plants having one contrasting character or a pair of contrasting characters. When he crossed pea plants having round and yellow seeds with pea plants having wrinkled and green seeds, he observed that no plants with wrinkled and green seeds were obtained in the F_1 generation. When the F_1 generation pea plants were cross-bred by self-pollination, the F_2 generation had seeds with different combinations of shape and colour also.
- (a) Write any two pairs of contrasting characteristics of pea plant used by Mendel other than those mentioned above. 1
 - (b) Differentiate between dominant and recessive traits. 1
 - (c) State the ratio of the combinations observed in the seeds of F_2 generation (in the above case). What do you interpret from this result ? 2

OR

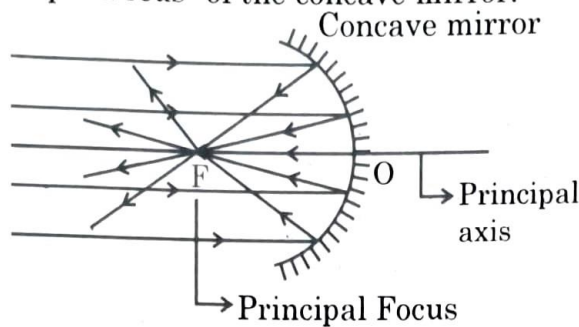
- (c) Given below is a cross between a pure violet flowered pea plant (V) and a pure white flowered pea plant (v). Diagrammatically explain what type of progeny is obtained in F_1 generation and F_2 generation :
 Pure violet flowered plant \times Pure white flowered plant.

(V V)

(v v)

2

38. Hold a concave mirror in your hand and direct its reflecting surface towards the sun. Direct the light reflected by the mirror on to a white card-board held close to the mirror. Move the card-board back and forth gradually until you find a bright, sharp spot of light on the board. This spot of light is the image of the sun on the sheet of paper; which is also termed as "Principal Focus" of the concave mirror.



- (a) List two applications of concave mirror. 1
- (b) If the distance between the mirror and the principal focus is 15 cm, find the radius of curvature of the mirror. 1
- (c) Draw a ray diagram to show the type of image formed when an object is placed between pole and focus of a concave mirror. 2

OR

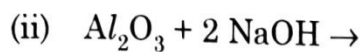
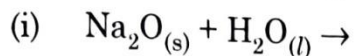
- (c) An object 10 cm in size is placed at 100 cm in front of a concave mirror. If its image is formed at the same point where the object is located, find : 2
- (i) focal length of the mirror, and
- (ii) magnification of the image formed with sign as per Cartesian sign convention.

39. Almost all metals combine with oxygen to form metal oxides. Metal oxides are generally basic in nature. But some metal oxides show both basic as well as acidic behaviour. Different metals show different reactivities towards oxygen. Some react vigorously while some do not react at all.

(a) What happens when copper is heated in air ? (Give the equation of the reaction involved). 1

(b) Why are some metal oxides categorized as amphoteric ? Give one example. 1

(c) Complete the following equations : 2



OR

(c) On burning Sulphur in oxygen a colourless gas is produced. 2

(i) Write chemical equation for the reaction.

(ii) Name the gas formed.

(iii) State the nature of the gas.

(iv) What will be the action of this on a dry litmus paper ?